

Selective oxidation of styrene to benzaldehyde using aluminium doped nano barium hexaferrite, synthesized by sol-gel auto-combustion method

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Abstract: Magnetic materials are extensively used as crucial components for a large multiplicity of industrial and technological applications. In the modern catalysis research particularly in organic synthesis, reusability and retrieval of catalysts is a significant feature. Compared to the unadventurous methods of separation, magnetic separation has established as a highly efficient, easy and rapid separation method for catalysts from its reaction mass. Barium Hexaferrite nanoparticles are well-known material, recognized as a hard ferrite with its preferred plane of magnetization perpendicular to the c-axis at room temperature, and these nanoparticles can be used as both catalyst and adaptable provision for functionalization of metals.

In the present work, the magnetic properties and catalytic properties of the aluminium doped Barium Hexaferrite nanoparticles are discussed. The Sol-gel auto-combustion technique was employed for the preparation of this nano-hexaferrite catalyst. The crystallinity and phase purity of synthesized ferrite nanoparticles were confirmed by the X-ray diffraction technique. The magnetic properties of the prepared ferrite nanoparticles were studied by SQUID based VSM technique at room temperature. The catalytic studies of the prepared ferrite catalyst were carried for the synthesis of Benzaldehyde, using styrene as a precursor in the presence of methanol: Water (2:3) and THF. These catalytic activities of the ferrite nanoparticles were recorded for this reaction and catalyst could be recovered easily by applying an external magnetic field.

Keywords: Ferrite, Catalyst, Dehydrogenation, Oxidation, Alkylation, THF

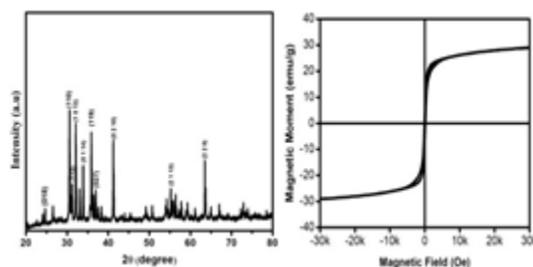


Figure 1: X-Ray diffraction pattern and M-H curve for aluminium doped barium hexaferrite synthesized via sol-gel autocombustion.

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