

Two-step novel synthesis of Mn-Ni-Co-Oxide(MNCO)@g-C₃N₄ ribbon-like hetero-nanoarrays for supercapacitor application

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Abstract: Since last few years, the hassle of clean, efficient, and renewable energy sources is remarkable and thereby driving intense scientific concern in the production, storage and management of this precious energy. Among variety of electrical energy storage devices, supercapacitors (SCs), also known as electrochemical capacitors, have industrial attention because of their high power density and very long cyclic stability as compared to their counterparts such as batteries, fuel cells and conventional capacitors etc.

To improve the electrochemical properties of the electrode materials, Mn-Ni-Co-Oxide (MNCO)@g-C₃N₄ Ribbon-like Hetero-nanoarrays were synthesized by a simple two-step approach including hydrothermal method and subsequent calcination process. This ternary metal oxide composite was directly utilized for the pseudocapacitive investigation because it is believed that these three metal elements can provide synergistic effects during the redox reaction process compared to the single component metal oxides. The chemical composition and morphology of the material was characterized by XRD, EDS and FESEM and HRTEM techniques. The electrochemical capacitance of MNCO@g-C₃N₄ was examined by cyclic voltammetry, galvanostatic charge–discharge and electrochemical impedance measurements. Results shows that the MNCO@g-C₃N₄ electrode material exhibits electrochemical activity for supercapacitor applications. A maximum specific capacitance value of 5624 F g⁻¹ could be obtained under an applied current density of 1 Ag⁻¹ in 6 mol L⁻¹ KOH electrolyte. It is envisaged that the novel MNCO@g-C₃N₄ ternary oxide composite material can be an alternate electrode material for high-performance supercapacitor application.

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